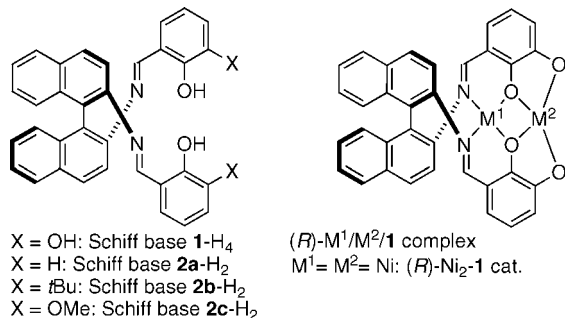


## Direct Catalytic Asymmetric Vinylogous Mannich-Type and Michael Reactions of an $\alpha,\beta$ -Unsaturated $\gamma$ -Butyrolactam under Dinuclear Nickel Catalysis

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Catalytic asymmetric vinylogous reactions of  $\gamma$ -butenolides and related compounds have been intensively studied, giving versatile functionalized chiral  $\gamma$ -butenolide skeletons.<sup>1–3</sup> In contrast, the use of their aza-analogues,  $\alpha,\beta$ -unsaturated  $\gamma$ -butyrolactams, as donors in catalytic asymmetric reactions is rare, despite their synthetic utility.<sup>1a,4</sup> For example, a vinylogous Mukaiyama Mannich-type adduct of a siloxypyrrole was utilized as a key building block for the synthesis of anti-influenza agent A-315675.<sup>4</sup> Lewis acid catalyzed asymmetric vinylogous Michael reaction of an acryloyloxazolidinone<sup>5</sup> and Lewis base catalyzed asymmetric vinylogous aldol reaction<sup>6</sup> using the siloxypyrrole have been recently reported. There are, however, no reports on catalytic asymmetric Mannich-type reactions of siloxypyrroles. Moreover, there is no example of direct catalytic asymmetric reactions of  $\alpha,\beta$ -unsaturated  $\gamma$ -butyrolactams under proton transfer conditions,<sup>7</sup> which are more favorable in terms of atom economy. Thus, the development of direct catalytic asymmetric reactions of  $\alpha,\beta$ -unsaturated  $\gamma$ -butyrolactams is highly desirable. Herein, we describe our efforts to address these issues. A homodinuclear Ni<sub>2</sub>-Schiff base **1** complex (Figure 1) promoted the direct catalytic asymmetric vinylogous Mannich-type reaction of an  $\alpha,\beta$ -unsaturated  $\gamma$ -butyrolactam with *N*-Boc imines and vinylogous Michael reaction to nitroalkenes, giving products in up to 99% ee in both reactions.



**Figure 1.** Structures of dinucleating Schiff base **1**-H<sub>4</sub>, bimetallic Ni<sub>2</sub>-Schiff base **1** complex, and Schiff bases **2**-H<sub>2</sub>.

To realize the direct vinylogous reaction with  $\alpha,\beta$ -unsaturated  $\gamma$ -butyrolactam **3**, the chemoselective activation of **3** as a donor and imine **4** as an electrophile is required. However, undesirable electrophilic activation of the  $\alpha,\beta$ -unsaturated  $\gamma$ -butyrolactam unit in both **3** and the vinylogous Mannich product must be avoided, as this may lead to side reactions, such as polymerization. In addition, the enantioselectivity, diastereoselectivity, and  $\alpha/\gamma$ -selectivity of the metal dienolate should be controlled. Among acid/base bifunctional catalysts developed in our group,<sup>8</sup> bimetallic Schiff base **1** catalysts,<sup>9–11</sup> which were previously used for catalytic deprotonation of the  $\alpha$ -proton in carbonyl donors, showed promising results in the present vinylogous reactions. The optimization studies using  $\alpha,\beta$ -unsaturated  $\gamma$ -butyrolactam **3** and *N*-Boc imine **4a** are sum-

marized in Table 1. Initial trials using dinuclear Mn<sub>2</sub>,<sup>9c</sup> Co<sub>2</sub>,<sup>9b</sup> Cu<sub>2</sub>, and Pd<sub>2</sub>-**1** complexes resulted in poor reactivity (entries 1–4). In contrast, a dinuclear Ni<sub>2</sub>-**1** complex<sup>10</sup> smoothly promoted the vinylogous reaction at room temperature. The  $\alpha$ -adduct was not detected at all, and the desired  $\gamma$ -adduct **5a** was obtained as a single isomer in 85% NMR conversion yield with 99% ee (entry 5). The yield was improved by the addition of DRIERITE (CaSO<sub>4</sub>) as a desiccant to give **5a** with >95% yield and 99% ee (entry 6). Molecular sieves also had beneficial effects to improve yield, but there was a reproducibility problem in terms of diastereoselectivity using molecular sieves. Thus, DRIERITE was utilized for further studies. Catalyst loading was successfully reduced to 5 mol % without loss of reactivity or stereoselectivity (entry 7). Control experiments using monometallic Ni-Schiff base **2a–2c** complexes did not afford the desired Mannich adduct **5a** but gave only byproducts (entries 8–10). In entries 11–12, heterobimetallic Pd/Ni-**1** and Cu/Ni-**1** complexes also did not give **5a**. The results in entries 8–12 suggested that two Ni metal centers are essential for the chemoselective 1:1 reaction of **3** with imine **4a**. We assume that a bimetallic Lewis acid/Brønsted base bifunctional mechanism would be operative in the present reaction, as we previously observed in other reactions.<sup>9,10</sup>

**Table 1.** Optimization of Reaction Conditions

entry	M <sup>1</sup>	M <sup>2</sup>	Schiff base	x	additive	time (h)	% yield <sup>a</sup>	dr <sup>a</sup>	% ee <sup>b</sup>
1	Mn-OAc	Mn-OAc	<b>1</b>	10	none	17	0	—	—
2	Co-OAc	Co-OAc	<b>1</b>	10	none	17	15	—	—
3	Cu	Cu	<b>1</b>	10	none	17	trace	—	—
4	Pd	Pd	<b>1</b>	10	none	17	3	—	—
5	Ni	Ni	<b>1</b>	10	none	17	85	>30:1	99
6	Ni	Ni	<b>1</b>	10	DRIERITE	24	>95	>30:1	99
7	Ni	Ni	<b>1</b>	5	DRIERITE	24	95	>30:1	99
8	Ni	none	<b>2a</b>	5	DRIERITE	24	0	—	—
9	Ni	none	<b>2b</b>	5	DRIERITE	24	0	—	—
10	Ni	none	<b>2c</b>	5	DRIERITE	24	0	—	—
11	Pd	Ni	<b>1</b>	5	DRIERITE	24	0	—	—
12	Cu	Ni	<b>1</b>	5	DRIERITE	24	0	—	—

<sup>a</sup> Determined by <sup>1</sup>H NMR analysis of crude mixture. <sup>b</sup> Determined by HPLC analysis using chiral column IA.

The substrate scope of the reaction is summarized in Table 2.<sup>12</sup> The Ni<sub>2</sub>-**1** catalyst was applicable to nonisomerizable aryl and heteroaryl imines. High enantio- and diastereoselectivity were achieved for aryl imines with either an electron-withdrawing or electron-donating substituent at the *ortho*-, *meta*-, or *para*-position (entries 2–7, 23:1–30:1 dr, 99% ee). With heteroaryl imines, the reactivity somewhat decreased. Thus, the reaction was performed

**Table 2.** Direct Catalytic Asymmetric Vinylogous Mannich-Type Reaction of an  $\alpha,\beta$ -Unsaturated  $\gamma$ -Butyrolactam and *N*-Boc Imines<sup>a</sup>

entry	R: 4	product 5	% yield <sup>b</sup>	dr <sup>c</sup>	% ee <sup>d</sup>	
1	Ph-	<b>4a</b>	<b>5a</b>	95	>30:1	99
2	2-naphthyl	<b>4b</b>	<b>5b</b>	76	30:1	99
3	1-naphthyl	<b>4c</b>	<b>5c</b>	93	26:1	99
4	2-Cl-C <sub>6</sub> H <sub>4</sub> -	<b>4d</b>	<b>5d</b>	95	>30:1	99
5	3-Me-C <sub>6</sub> H <sub>4</sub> -	<b>4e</b>	<b>5e</b>	87	>30:1	99
6	4-Cl-C <sub>6</sub> H <sub>4</sub> -	<b>4f</b>	<b>5f</b>	87	>30:1	99
7	4-MeO-C <sub>6</sub> H <sub>4</sub> -	<b>4g</b>	<b>5g</b>	85	23:1	99
8 <sup>e</sup>	2-furyl	<b>4h</b>	<b>5h</b>	61	5:1	99
9 <sup>e</sup>	3-thienyl	<b>4i</b>	<b>5i</b>	83	21:1	99

<sup>a</sup> Reaction was run using 1 equiv of **3**, 1.2 equiv of **4**, in THF (0.3 M) in the presence of DRIERITE unless otherwise noted. <sup>b</sup> Isolated yield after purification by column chromatography. <sup>c</sup> Determined by <sup>1</sup>H NMR analysis of crude mixture. <sup>d</sup> Determined by HPLC using chiral column IA or AD-H. <sup>e</sup> Reaction was run using 2 equiv of **3** and 1 equiv of **4**.

**Table 3.** Direct Catalytic Asymmetric Vinylogous Michael Reaction of an  $\alpha,\beta$ -Unsaturated  $\gamma$ -Butyrolactam to Nitroalkenes<sup>a</sup>

entry	R: 6	cat. (x mol %)	time (h)	7	% yield <sup>b</sup>	dr <sup>c</sup>	% ee <sup>d</sup>	
1	Ph-	<b>6a</b>	2.5	13	<b>7a</b>	98	>30:1	97
2	4-Br-C <sub>6</sub> H <sub>4</sub> -	<b>6b</b>	2.5	12	<b>7b</b>	96	29:1	99
3	2-Br-C <sub>6</sub> H <sub>4</sub> -	<b>6c</b>	2.5	11	<b>7c</b>	98	16:1	98
4	4-MeO-C <sub>6</sub> H <sub>4</sub> -	<b>6d</b>	2.5	25	<b>7d</b>	98	>30:1	98
5	2-furyl	<b>6e</b>	2.5	11	<b>7e</b>	98	>30:1	98
6	2-thienyl	<b>6f</b>	2.5	15	<b>7f</b>	99	>30:1	98
7	PhCH <sub>2</sub> CH <sub>2</sub> -	<b>6g</b>	2.5	24	<b>7g</b>	89	26:1	96
8	<i>i</i> -propyl	<b>6h</b>	2.5	24	<b>7h</b>	97	>30:1	99
9	( <i>E</i> )-Ph-CH=CH-	<b>6i</b>	2.5	16	<b>7i</b>	83	25:1	99
10	Ph-	<b>6a</b>	1	36	<b>7a</b>	84	29:1	93

<sup>a</sup> (*S*)-Ni<sub>2</sub>-1 catalyst was used in Table 3. Reaction was run using 2 equiv of **3** in 1,4-dioxane (0.15 M) at 50 °C. <sup>b</sup> Isolated yield after purification by column chromatography. <sup>c</sup> Determined by <sup>1</sup>H NMR analysis of crude mixture. <sup>d</sup> Determined by HPLC using chiral column IC, IB, or AD-H.

with 2 equiv of **3**, and products **4h–4i** were obtained in 99% ee (entries 8–9). Unfortunately, isomerizable aliphatic imines resulted in low yield (<20%) due to competitive isomerization to enamides over nucleophilic activation of **3**.

Trials to further expand the vinylogous nucleophilicity of **3** under bimetallic Schiff base catalysis revealed that vinylogous Michael reaction to nitroalkenes<sup>3b</sup> proceeded nicely in 1,4-dioxane at 50 °C using the same Ni<sub>2</sub>-1 catalyst. As summarized in Table 3, the reaction of aryl, heteroaryl, and alkyl substituted nitroalkenes **6a–6h** proceeded smoothly with 2.5 mol % catalyst loading, and products were obtained in 89–99% yield, 16:1–30:1 dr, and 96–99% ee after 11–25 h (entries 1–8).<sup>12</sup> Nitrodiene **6i** was also applicable, and the 1,4-adduct was predominantly obtained in 83% yield, 25:1 dr, and 99% ee (entry 9). Catalyst loading was successfully reduced to 1 mol %, although enantioselectivity decreased to 93% ee (entry 10).

In summary, we developed direct catalytic asymmetric vinylogous Mannich-type and Michael reactions of  $\alpha,\beta$ -unsaturated

$\gamma$ -butyrolactam. Negative control experiments clearly suggested the importance of a dinuclear Ni system. The dinuclear Ni-catalyzed reactions proceeded selectively at the  $\gamma$ -position, giving vinylogous Mannich adducts in 5:1–30:1 dr and 99% ee and vinylogous Michael adducts in 16:1–30:1 dr and 93–99% ee. Further studies to expand the vinylogous nucleophilicity under dinuclear Schiff base catalysis, including trials to improve the yield with enolizable aliphatic imines, are ongoing.

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**Supporting Information Available:** Experimental procedures, spectral data of new compounds, and cif files. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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